**Fractions** 

$$\frac{a}{b}$$
 means  $a \div b$ 

$$\frac{a}{b}$$
 means  $a \div b$  a is the same as  $\frac{a}{1}$ 

**Equality of Fractions** 

If 
$$\frac{a}{b} = \frac{c}{d}$$
, then  $a \times d = b \times c$ 

(cross multiplication; equivalent to multiplying both sides by bd)

**Multiplying Fractions** 

$$\frac{a}{b} \times \frac{c}{d} = \frac{a \times c}{b \times d}$$

**Dividing Fractions** 

$$\frac{a}{b} \div \frac{c}{d} \equiv \frac{a}{b} \times \frac{d}{c} = \frac{a \times d}{b \times c} \quad (\text{same as } \frac{\frac{a}{b}}{c} = \frac{a \times d}{b \times c})$$

## **Scientific Notation**

The basis of all science is measurement. Using very large or very small numbers becomes cumbersome. In these cases, we rewrite the numbers in scientific notation (a.bcd x 10<sup>e</sup>). For example, 342587 becomes  $3.42587 \times 10^5$  and 0.000864 can be written as  $8.64 \times 10^{-4}$ .

**Multiplying Numbers in Scientific Notation** 

multiply numbers and add exponents

$$(a \times 10^{x}) \times (b \times 10^{y}) = ab \times 10^{x+y}$$

**Dividing Numbers in Scientific Notation** 

divide numbers and subtract exponents

$$(a \times 10^{x}) \div (b \times 10^{y}) = \frac{a}{b} \times 10^{x-y}$$

**Percent** (%) = (parts) per hundred

In a basket of 40 apples, 10 are rotten; % rotten apples =  $\frac{10}{40}$  x 100 = 25%

ppm = parts per million; ppb = parts per billion

A 555,000 g sample of water contains 35 g of a toxic substance; express the amount of the toxic substance in ppm.

ppm = 
$$\frac{35 \text{ g toxic substance}}{555,000 \text{ g water}} \times 1,000,000 = 63 \text{ ppm}$$

## Find the error in the story below.

Vancouver Sun, Aug. 24, 1987, B1

Brodeur (Canuck's No.1 goaltender for the last 7 years) was offered a new one-year (plus option) contract that includes both a pay cut at the major league level and a minor league clause. The latter would, according to Watters (Brodeur's agent), mean a 200 per cent pay cut if Brodeur was sent to the minors. Brodeur is reportedly paid \$300,000 per season.

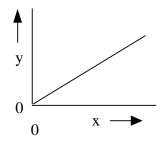
## **Direct Proportionality**

For any material, the mass and volume are directly proportional; if the volume is doubled, the mass doubles, etc.

$$M \propto V$$
 or  $M = kV$  (k is a onstant)

 $\frac{M}{V} = k$  (if two variables are directly proportional, the *ratio* is constant)

usual form is y = kx which is the equation for a straight line graph through the origin (0,0)



## **Inverse Proportionality**

For a sample of a gas at constant temperature, the volume is inversely proportional to the pressure; if the pressure is doubled, the volume is halved, etc.

$$V \propto \frac{1}{P}$$
 or  $V = k \times \frac{1}{P}$ 

or PV = k (if two variables are inversely proportional, the *product* is constant)

If xy = k, a graph of y vs x is a curve, but a graph of y vs 1/x is a straight line

